

# GCSE

## **Additional Science B**

General Certificate of Secondary Education

Unit B623/02: Modules B3, C3, P3 (Higher Tier)

### Mark Scheme for June 2012

OCR (Oxford Cambridge and RSA) is a leading UK awarding body, providing a wide range of qualifications to meet the needs of candidates of all ages and abilities. OCR qualifications include AS/A Levels, Diplomas, GCSEs, OCR Nationals, Functional Skills, Key Skills, Entry Level qualifications, NVQs and vocational qualifications in areas such as IT, business, languages, teaching/training, administration and secretarial skills.

It is also responsible for developing new specifications to meet national requirements and the needs of students and teachers. OCR is a not-for-profit organisation; any surplus made is invested back into the establishment to help towards the development of qualifications and support, which keep pace with the changing needs of today's society.

This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which marks were awarded by examiners. It does not indicate the details of the discussions which took place at an examiners' meeting before marking commenced.

All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

OCR will not enter into any discussion or correspondence in connection with this mark scheme.

© OCR 2012

Any enquiries about publications should be addressed to:

OCR Publications PO Box 5050 Annesley NOTTINGHAM NG15 0DL

Telephone:0870 770 6622Facsimile:01223 552610E-mail:publications@ocr.org.uk

#### Annotations used in Scoris

Annotation	Meaning
✓	Correct response
×	Incorrect response
HOD	Benefit of the doubt
<b>1800</b>	Benefit of the doubt <u>not</u> given
	Error carried forward
<b>^</b>	Information omitted
I	Ignore
	Reject
(H+).(	Contradiction

#### Subject-specific Marking Instructions

Abbreviations, annotations and conventions used in the detailed Mark Scheme.

- = alternative and acceptable answers for the same marking point
- (1) = separates marking points
- **allow** = answers that can be accepted
- **not** = answers which are not worthy of credit
- reject = answers which are not worthy of credit
- **ignore** = statements which are irrelevant
- () = words which are not essential to gain credit
  - \_ = underlined words must be present in answer to score a mark (although not correctly spelt unless otherwise stated)
- ecf = error carried forward
- AW = alternative wording
- ora = or reverse argument

Qı	Question		Answer		Guidance
1	(a)		diploid	2	each incorrect tick loses 1 mark down to zero
			haploid		
			meiosis		
			mitosis		
			multiplication		
	(b)		idea that humans tend to grow to a finite size but (many) plants can grow continuously / plants have meristems and grow at the tips (1)	1	<b>allow</b> humans stop growing (at adulthood) but plants don't <b>allow</b> humans have proportional growth <b>allow</b> plant growth is seasonal
	(c)		larger animals have to have longer period (for cell division to produce functional organs) / ora (1)	1	ignore it depends on the size of the animal ignore references to complexity or numbers of cells
	(d)		able to take on new roles / AW / able to become specialised /	1	allow not specialised
			able to develop into different kinds of cells (1)		allow can turn into any cell
	(e)		(multi-cellular microorganisms are small and) have larger surface area to volume ratio / ora (1)	1	allow surface area / volume ratio but ignore surface area unqualified ignore organism can be larger
					ignore so cells can differentiate ignore so organism can be more complex
			Total	6	

Q	Question		Answer	Marks	Guidance
2	(a)		build up to form a plaque (1)	2	<ul> <li>not veins / capillaries (penalise only once)</li> <li>allow cholesterol sticks to walls or lining or inside of arteries</li> <li>allow cholesterol builds up or collect in arteries or (blood) vessels</li> <li>allow fat for cholesterol</li> <li>not cell walls</li> </ul>
			idea that blood flow is restricted or blocked (in arteries) (1)		<b>allow</b> slow down blood flow / restrict circulation of blood <b>allow</b> blocks or clogs up arteries or (blood) vessels / makes arteries narrower / AW
	(b)		$\frac{(40 \times 8.5 = ) 3.4 \text{ (mmol)} (1)}{(100)}$ <b>but</b>	3	allow ecf over the 3 steps
			(8.5 – 3.4 =) 5.1 (mmol) (2) <b>but</b> (5.7 – 5.1 =) 0.6 (mmol) (3) <b>but</b> 0.6 (mmol per litre of blood) on its own (3)		<b>allow</b> $\frac{60 \times 8.5}{100} = 5.1 (2)$
	(C)		idea of movement (of a substance or solute) from a high to a low concentration (1)	1	<b>allow</b> goes from a high concentration to a low concentration (1) <b>ignore</b> breaking down from a high to a low concentration <b>ignore</b> references to membranes <b>not</b> movement of cells
	(d)		any two from: long / large surface area (1) villi / microvilli (1)	2	
			permeable or thin cell surface or cell membrane or lining (1)		allow lining one cell thick allow (intestine) walls are thin but not cell walls are thin
			good blood supply / many capillaries (1)		ignore close to the blood
			Total	8	

Question		on	Answer	Marks	Guidance
3	(a)	(i)	insert (1)	2	
			resistance (1)		
		(ii)	plant cells retain ability to differentiate (1)	2	allow plant cells can turn into different cells throughout their life
			animal cells usually lose this ability (at an early stage) (1)		<b>allow</b> stem cells found in developed animals have limited differentiation
	(b)		any two from:	2	
			base sequence or base code determines amino acid sequence (1)		
			each amino acid is coded for by (a sequence of) 3 bases (1)		
			<b>sequence</b> of amino acids makes up a protein (1)		
			Total	6	

Q	Question		Answer	Marks	Guidance
4	(a)	(i)	proton	1	both required for mark ignore positive and neutral
		(ii)	12 / twelve (1)	1	
	(b)	(i)	<b>C</b> (1)	1	<b>allow</b> correct answer ticked, circled or underlined in list if answer line is blank
		(ii)	covalent (1)	1	ignore strong
	(c)		(low melting point) because it has weak intermolecular forces (1) (does not conduct electricity) because it does	2	<ul> <li>allow forces / bonds / attraction between molecules for intermolecular forces</li> <li>not weak metallic bonding not forces between atoms / intramolecular forces must be clear that forces are between molecules e.g. forces holding molecule together scores 0</li> <li>ignore because it is not a metal</li> </ul>
			not have any free electrons / no delocalised electrons (1)	6	not reference to ions
			TOLAI	U	

Q	Question		Answer	Marks	Guidance
5	(a)		all have one electron in their outer shell (1)	1	allow all lose one electron allow they will have the same number of electrons in the outer shell
	(b)		2Na + 2H <sub>2</sub> O → 2NaOH + H <sub>2</sub> correct reactants and products (1) correct balancing (1)	2	<ul> <li>allow any correct multiple, including fractions</li> <li>allow = / ⇒ instead of →</li> <li>not and / &amp; instead of +</li> <li>balancing mark is dependent on the correct formulae but</li> <li>allow 1 mark for a balanced equation with a minor error in subscripts / formulae</li> <li>e.g. 2Na + 2H2O → 2NaOH + H2</li> </ul>
	(c)		(potassium) loses outer electron more easily (than sodium) (1)	1	<ul> <li>allow reverse argument for sodium</li> <li>allow potassium has more (shielding) shells / outer shell is further from the nucleus / outer electron is further from the nucleus but potassium has a bigger atom is not sufficient</li> <li>allow potassium has a weaker force between the nucleus and the outer electron</li> <li>ignore (potassium) loses outer electron more quickly ignore potassium has more electrons</li> <li>assume unqualified answer refers to potassium</li> </ul>
			Total	4	

Q	Question		Answer	Marks	Guidance
6	(a)		state of chlorine – gas (1) colour of bromine – red / brown / orange / yellow (1) radius of astatine – any value between 0.160 and 0.200 (1)	3	<b>allow</b> any combination of red, brown, orange and yellow (1) e.g. orange-red or red-brown <b>allow</b> rusty red or foxy red (1) <b>not</b> combinations where one of the colours is incorrect e.g. black-brown <b>ignore</b> references to pale or dark
	(b)	(i)	bromine + sodium iodide $\rightarrow$ iodine + sodium bromide (1)	1	allow = instead of → not and / & / instead of + allow correct formulae (i.e. case and subscripts must be correct) but equation does not need to balance e.g. Br <sub>2</sub> + Nal → l <sub>2</sub> + NaBr (1) allow mix of correct formulae and words not Br + Nal → I + NaBr
		(ii)	$Cl_2 + 2e^- \rightarrow 2Cl^-$ correct reactants and products (1) correct balancing (1)	2	allow $Cl_2 \rightarrow 2Cl^ 2e^-$ allow any correct multiple, including fractions allow = / = instead of $\rightarrow$ not and / & instead of + balancing mark is dependent on the correct formulae but allow 1 mark for a balanced equation with a minor error in subscripts / formulae e.g. $Cl_2 + 2e^- \rightarrow 2Cl^-$ allow $Cl + e^- \rightarrow Cl^-$ or $Cl \rightarrow Cl^ e^-(1)$
			Total	6	

Q	Question		Answer	Marks	Guidance
7	(a)		idea that copper (still) has a high (electrical) conductivity / idea that (electrical) conductivity of copper is similar to silver(1) idea that copper is cheaper / ora (1)	2	allow density of copper is less than silver (1) ignore any comments about corrosion ignore wires are heavy
					allow because of density and cost (1) if no other mark scored
	(b)		(good) conductor of heat / (good) thermal conductor (1)	1	allow malleable allow does not corrode / does not rust allow it is lustrous ignore properties from the table
	(c)		metals have strong metallic bonds (1)	1	
			Total	4	

Q	Question		Answer	Marks	Guidance
8	(a)		E (1)	1	more than one letter scores zero. if answer line is blank allow correct answer indicated on list
	(b)		600 (m) (1)	1	
	(c)		4500 (N) (3) <b>but if answer incorrect</b> correct calculation of acceleration as $5(m/s^2)$ (2) <b>if incorrect</b> $\frac{20}{4}$ (1) <b>or</b> (force =) calculated acceleration x 900 (1)	3	allow ecf for incorrectly calculated acceleration e.g. a = 4/20 = 0.25 then F = 900 x 0.25 = 225 scores 1 allow (F =) ma or (force =) mass x acceleration (1) not 900 x 20 or 900 x 4
	(d)		area under graph (1)	1	<b>allow</b> correct area calculation $\frac{1}{2} \times 20 \times 4 = 40$ (1) <b>allow</b> multiply the speed by 4 and then half it (1)
			Total	6	

Q	Question		Answer		Guidance
9	(a)		$ \begin{array}{c c} \hline & \checkmark \\ \hline & & \\ \hline \\ \hline & \\ \hline & \\ \hline \\ \hline \\ \hline \\ \hline \\ \hline \hline \\ \hline \\ \hline \hline \hline \\ \hline \hline \hline \hline \\ \hline \hline \hline \hline \hline \hline \\ \hline \hline$	2	all correct (2) any two horizontal lines correct (1)
	(b)		<ul> <li>any two from:</li> <li>car B brakes after A / ora (1)</li> <li>idea that car B has not left or allowed enough braking distance / thinking distance (1)</li> <li>idea that distance between the cars is shorter than the stopping distance ora (1)</li> <li>braking (distance) starts when graph(s) curves or car(s) have stopped when lines are horizontal / flat (1)</li> <li>B overtakes or goes past A where the graphs</li> </ul>	2	allow car <b>B</b> driving within the braking distance of <b>A</b> (1) allow car <b>B</b> does not start to break until it has nearly caught up with car <b>A</b> (1) ignore car <b>B</b> is driving too close to car <b>A</b> allow cars would collide where graphs cross (1)
			cross (1) the graph for <b>B</b> should finish up under or below the graph for <b>A</b> (1)		allow the <b>B</b> line finishes above the <b>A</b> line (1)
			Total	4	

Q	Question		Answer	Marks	Guidance
10	(a)			3	
			concrete barrier (no mark) because stopping distance is small / smaller / smallest or time (to stop) is short / shorter / shortest (1)		no marks if escape lane or metal crash barrier is chosen or no safety feature chosen
			(so) acceleration is large / larger / largest (1) (so) force is large / larger / largest (1)		<b>allow</b> deceleration for acceleration
			idea that large / larger / largest acceleration <b>or</b> force happens (when hitting the concrete barrier as it is) when time or distance is small / smaller / smallest / ora (2)		
	(b)		(if force is less) so deceleration is less (1)	2	allow deceleration takes longer (1) allow acceleration for deceleration (1)
			(hence) time to stop or stopping distance increases (1)		allow it takes longer for the car to stop (1) allow any of the following equations in any correct rearranged form f = ma(1) $a = \frac{change in speed}{time taken}$ or $speed = \frac{distance}{time taken}$ (1)
			Total	5	

Q	Question		Answer	Marks	Guidance
11	(a)		arrow on ball pointing downwards <b>and</b> labelled gravity / weight (1) arrow on ball pointing upwards <b>and</b> labelled drag / <b>air</b> resistance / <b>air</b> friction / <b>wind</b> resistance (1)	2	drag / air resistance gravity / weight ignore upthrust
	(b)	(i)	forces are balanced (1)	1	<ul> <li>allow weight = drag</li> <li>allow equal and opposite</li> <li>allow equal if forces correctly drawn and labelled in part (a)</li> <li>not equal if forces incorrectly drawn and labelled in part (a)</li> <li>if named forces used allow ecf from Q11(a)</li> </ul>
		(ii)	(gravitational potential energy) decreases (steadily) (1) (kinetic energy) increases (steadily) until terminal velocity <b>or</b> (kinetic energy) increases (steadily) and then remains constant (1)	2	<ul> <li>ignore changes to gravitational potential energy once it hits the ground</li> <li>ignore changes to kinetic energy once it hits the ground</li> </ul>
			Total	5	

OCR (Oxford Cambridge and RSA Examinations) 1 Hills Road Cambridge CB1 2EU

**OCR Customer Contact Centre** 

#### **Education and Learning**

Telephone: 01223 553998 Facsimile: 01223 552627 Email: general.qualifications@ocr.org.uk

#### www.ocr.org.uk

For staff training purposes and as part of our quality assurance programme your call may be recorded or monitored

Oxford Cambridge and RSA Examinations is a Company Limited by Guarantee Registered in England Registered Office; 1 Hills Road, Cambridge, CB1 2EU Registered Company Number: 3484466 OCR is an exempt Charity

OCR (Oxford Cambridge and RSA Examinations) Head office Telephone: 01223 552552 Facsimile: 01223 552553



